

Assignment - 1  
B.Sc. 4<sup>th</sup> Sem

Marks = 10

- Q. 1) What is the reason for increased atomic vol<sup>m</sup> after grp. 10.  $\equiv$
- Q. 2) How can you explain the low melting pts of Zn, Cd & Hg in comparison to other d-block elements.  $\equiv$
- Q. 3) Name the important oxidation states of iron. Why is Fe(III) more stable than Co(III) and Ni(III). ?  $1+2\equiv 3$
- Q. 4) Why d-block elements are not as good reducing agents as the metals of Grp. 1, 2 & 13 ?  $\equiv$
- Q. 5) Calculate in Bohr Magnetons the expected magnetic moment for Mn<sup>2+</sup> ion. ( $\mu$  spin only)  $\equiv$